## Chemistry 1A Greensheet



Requests for review of exam or quiz: If you believe that I have added points incorrectly or did not provide sufficient credit, then follow the procedure below:

1) Thoroughly review the key!
2) Attach a separate piece of paper to the top of the quiz or exam
3) Describe what I need to review. i.e. did I total points incorrectly? should additional partial credit be provided (specify the question \#) and why, etc.
4) Turn in the exam/quiz within one week (but not the day I returned it).
5) After one week, no adjustments will be made.

Remember, I may provide additional points if warranted. However, if your answer is full of incorrect statements or I find additional problems, I may also take off more points if warranted. So, please review your answer and the key completely.

Cheating: The minimum penalty for cheating on an exam, or plagiarism in the lab, is the assignment of a zero on the assignment in question. The matter will be referred to the DeAnza administration for appropriate action and possible further discipline. YOU are responsible for understanding the De Anza Academic Integrity policy

Attendance: I may drop any individual that is not present at the first scheduled class meeting or is 15 minutes late to the second class meeting. It is your responsibility to insure that you have properly dropped this course. Your work load,
course load, transportation difficulties are all avoidable! The message: You must be academically prepared and be committed to this class. The failure rate for this class is typically approximately $\mathbf{3 0 \%}$. The common reasons are 1) lack of academic preparation (usually poor algebra skills), 2) lack of study time, or 3) too heavy a course load.

It will be rare (hopefully not at all) that I arrive late for class. I expect the same from you.

Miscellaneous: Cellular phones must be turned off and put away during lecture . ONLY NON-PROGRAMMABLE calculators are allowed during quizzes and examinations. That is, the TI 84/85 series or similar calculators MAY NOT be used. As we regularly have quizzes or exams, I strongly recommend that you always bring your calculator with you.

## Student Learning Outcome(s):

${ }^{*}$ Identify and explain trends in the periodic table.
${ }^{*}$ Construct balanced reaction equations and illustrate principles of stoichiometry.
*Apply the first law of thermodynamics to chemical reactions.

