Chemistry 25: Prep for General Chemistry

 Lecture:
 T/Th
 8:30 - 10:20 AM
 SC 22997/22998

 Lab-1
 T
 11:30 AM - 12:20 PM
 SC 22997

 Lab-2
 Th
 11:30 AM - 12:20 PM
 SC 22998

Instructor: Dr. Rachel Snelling **Phone:** N/A **E-mail:** snellingrachel@fhda.edu

Office Hours: TBA Office: Zoom

This course syllabus is a contract:

One purpose of this syllabus is to provide you with the guiding principles upon which the class runs. Another purpose is to make sure that you have, at your fingertips, answers to common questions that might arise. This document is available at all times on the class website. Make sure you read it in its entirety before you ask me any questions about the course schedule, requirements, grading, etc... It is also a contract between you the student, and I, the instructor of record. Make sure that you understand its contents fully, especially the parts that pertain to testing and the computation of your grade, because so long as you remain enrolled in the course, you are implicitly agreeing to abide by these terms.

Course Description: Chemistry 25 is meant to serve as an introduction to and grounding in the core theory and problem-solving techniques of chemistry as a preparation for a General Chemistry course (Chem 1A) and other science related fields. Conceptual topics include modern atomic and molecular theory, the mole and stoichiometry, behavior of gases, thermochemistry, and an exploration of the standard classes of chemical reactions. Laboratory topics covered include an introduction to gravimetric and volumetric analysis, introductory lab equipment and techniques, and keeping a laboratory notebook. Throughout all topics we will stress both conceptual and mathematical problem-solving techniques in order to prepare students to tackle these topics more in depth in following classes.

Prerequisites: MATH 114A or equivalent

Course Materials (Required):

- 1. **Text Book**: *Introduction the Chemistry*, 5e, by Bauer, Birk, Marks (McGraw Hill: 2019; ISBN 978-1-307-23515-9)
- 2. Scientific Calculator

Course Materials (Optional):

Resources

Tutoring: De Anza's tutorial center is in S43. This and many other campus services can be found as part of the student success center: http://www.deanza.edu/studentsuccess

Disability Support Program and Services: DSPS can help you get the right tools to succeed. Their website is http://www.deanza.edu/dsps/

Grading Scheme:	Percentage
Laboratory Work	20
Chapter Exams (3)	60
Final Exam	20
Total	100%

Laboratory Work (20%): You will be expected to participate in lab, complete lab reports, and pass lab quizzes. More details on these items can be found on the laboratory section.

Chapter Exams (60%): There will be 3 chapter exams worth 20% of your grade each. Exams will be a combination of any of the following: multiple choice, short synthesis or mechanisms, and vocabulary questions. Early and late exams are not administered. Missing an exam **will result in a zero** for that exam without proof of an excused absence (doctor's note, police report, etc...)

Final Exam (20%): The Final Exam is cumulative and will have the same format as the chapter exams. The exam will be given Thursday, December 10th from 7:00 AM – 9:00 AM. If you cannot make this time, you should not enroll in this class.

Special Note: If your average percentage is failing (<55%) in any **ONE** or more of the following portions of the course, you will not receive a passing grade:

exams or lab reports/assignments. Additionally, you must complete **all**_of the lab experiments/reports in order to pass the class.

Class Policies.

- **A. Time Requirement:** This class includes appx. 4 hours of lecture and appx. 3 hours of lab per week. In order to receive a "C" or better grade, you should allow 8-12 hours of studying, reading, and preparing outside of class **PER WEEK**. Help yourself to do your best by making time to keep up with the reading and homework. If this time commitment is not possible given your current situation, please consider taking this class at a later date when you do have more time available.
- **B.** Lecture Attendance: Attendance is a critical component of the learning process, and the lecture will cover material that may not appear in your text and help clarify the material that is. Learning Chemistry effectively depends on building up from a base of knowledge. If you do not set a firm foundation, you will not be able to build your understanding of the field effectively. In other words, miss too many classes and you'll likely fail the class.
- C. Class Behavior: Be ready to start class at the scheduled time. Please arrive on time and plan on staying the entire session as late arrivals and early departures distract everyone. If you are unavoidably late, please enter quietly and find your seat as quickly and quietly. Please do not disrupt class with irrelevant conversations, either in the form of inappropriate comments or private conversations. I would always prefer you show up a little late as opposed to skipping the class entirely.

- **D.** Please turn OFF your cell phone when you enter the class or lab. You may NOT take calls or texts during either, except for emergencies. Students caught abusing this rule may be docked points or expelled from class or lab.
- **E.** Academic Dishonesty: Cheating or plagiarizing another student's work, in whole or part, will result in a zero for the assignment, a referral to the dean and my immense displeasure. Any case where you attempt to gain unfair advantage over other students or attempt to pass off another's work as your own **is cheating**. Please see me if you have any questions. You implicitly agree to abide by the Honor Code as a condition of enrollment in this class:
- **F. Grading:** This class is not graded on a curve. Grade cut offs are as follows: A+ (97), A (93), A- (90), B+ (87), B (83), B- (80), C+ (76), C (69), D+ (65), D (60), D- (56), F (56-0)

http://www.deanza.edu/studenthandbook/academic-integrity.html

- **G. Extra Credit:** Extra credit assignments are not offered in this class on an individual basis. It is unfair to allow some students to improve their grade while not allowing others that same opportunity. Some extra credit problems may appear at the end of exams and in homework.
- **H. Dropping the Class:** If you wish to drop the class after the first 2 weeks, it is your responsibility to do so. If you fail to drop the class you will be assigned a grade in keeping with your submitted work, usually an F.
- **I. Questions/Help:** I am available to answer questions during office hours, by email, or by appointment. Please feel free to contact me with any problems or concerns that you have. Also remember that your fellow students are great resources.

Attendance Note

You are responsible for all the material covered in this course, and it is expected that you attend and participate in all of the lecture and laboratory sessions. *If you must be absent, then it is in your best interest to contact your instructor as soon as possible in order to find out what work you have missed.***Due to the high number of students wishing to enroll in this class, any unjustified absences during the **first two weeks of class will result in you being dropped.

LAB POLICIES:

LABORATORY CHECK-IN

Locker check-in will take place the first day of lab. It is your responsibility to make sure that all glassware is present and unbroken at the time you check in. If at any point after the first day of lab you need to replace an item in your locker, your student account will get charged for it. If you drop this course, then you must arrange to check-out your locker with your instructor during your regularly scheduled lab period. The stockroom technician or other instructors WILL NOT checkout lockers for any students. Any person who has not checked out by the end of the last scheduled lab period for the quarter will have an administrative fee added to their student account and a hold put on their registration.

LABORATORY PROCEDURES AND POLICIES

All students are expected to arrive to lab on time and to come to lab prepared to carry out the experiment scheduled for that session. This means that you have studied the experiment for the day, have a basic understanding of its purpose and procedure, the chemistry involved and *have prepared your laboratory notebook for the experiment prior to the start of lab*. I ask that all students do a conscientious and thorough job of cleaning up after themselves, whether it is in their own work area in the lab, or shared areas such as the chemical supply table and balance room.

LABORATORY SAFETY

Laboratory safety is an everyday assignment. *Being safe in the lab is a top priority*. The importance of safety in the laboratory will be reviewed the first day of lab. *Any unsafe behavior, intentional or not, will be noted and may be cause for dismissal from the class*.

For your protection, **safety goggles** with indirect ventilation and an ANSI minimum rating of Z87 **must be worn AT ALL TIMES** in the laboratory. **ONE warning** will be issued to any student that is observed wearing their goggles on their forehead, hanging them around their neck, etc... instead of wearing over their eyes. If the **warning is disregarded**, **expulsion** from the lab and a zero on the assignment may result.

LABORATORY LECTURE

The beginning of each laboratory session is designated as a laboratory lecture period for which you **must be on time** in order to perform the scheduled experiment. The instructor will use this lecture period to outline important details of the procedure, overview theory and calculations, and to emphasize safety hazards and proper chemical disposal. *If you are more than* **10** *minutes late for lab lecture, you will not be allowed to do the experiment for that day.*

ATTENDANCE

Attendance is required at all scheduled laboratory sessions. NEVER plan on missing a lab. *You will receive a zero on the second lab you miss and will fail the course on the third.* These absences include those in which you arrive too late for lab lecture and are thus not allowed to complete the experiment. I may allow for emergencies and other complications in life. Additionally, do not plan on leaving lab early. Labs will regularly take the total amount of time allotted.

CHEMICAL DISPOSAL

As a concern for the environment and to follow county, state and federal law, proper chemical disposal is essential. Students who do not comply with directed procedures may be expelled from the lab or failed in the course for repeated offenses. Check with the instructor if you have any questions.

LAB REPORTS

All lab reports must be completed and turned in to receive a passing grade in this class. Using another student's data or making up data is plagiarism and data falsification and will result in a zero for the assignment and referral to the dean. In cases where a student was unable to complete a lab, the instructor may direct you to use another's data in order to complete follow up quests at his discretion. The source of your data must always be cited in lab reports.

LATE ASSIGNMENTS

Due dates for assignments are listed on the class schedule. Late assignments will lose 20% of their value per period missed. Assignments in excess of 2 periods late will be graded as zeros (All Labs must still be turned in to receive course credit).

It is the student's responsibility to know when labs are due based on the provided class schedule. Labs are always due 2 lab periods (usually 1 week) after the lab session in which they are completed.

EXCUSED ABSENCE

Every student gets one excused absence. To reflect this, your lowest pre-lab and lab report are dropped at the end of the quarter. Missing a 2nd lab will result on a score of zero on that lab. **Missing a third will result in failing the course.**

Tips for Success

- Come to class having read the assigned chapter and be ready with questions about the concepts you didn't understand.
- In case you didn't read the first one, really, come to class with the assigned chapter already read. I cannot stress how big a difference this will make for you.
- Take notes during class and reread your notes before the next class. If something is still unclear, write down your question so you can ask about it during the next class or in office hours.
- Work every day. The longer the time that passes between doing chemistry problems, the more knowledge you have to rebuild. Do some homework problems and some problems from the book every day as this will help you understand where you need help, and it will help prepare you for the exams. Schedule some time each day to work on chemistry. Treat this subject like a foreign language. Use it or lose it.
- **Do the suggested chapter problems in the book**, particularly for concepts you're having trouble with.

- **Don't try to memorize EVERYTHING.** This is a common trap that many students fall into. While there are certain topics that must be committed to memory, strive to develop an intuitive understanding of the underlying framework of the material. Once you have that you will often be able to derive answers from a much smaller pool of "memorized" data.
- **Join a study group**, exchange phone numbers of classmates whom you can call for help. In the group, take time to present concepts to one another. The BEST way to solidify a topic in your mind is to have to teach it to someone else.
- **Don't wait** until the night before to finish that lab report or homework assignment. You'll get more out of it (and do better) if you give yourself the time to understand the concepts and ask questions when you get stuck.
- Start studying for the exams at least a week before. Cramming for an exam is like playing Russian Roulette! Cramming is superficial knowledge only, and when you are nervous, superficial knowledge is very unreliable.
- Work through old quizzes and homework problems before exams.
- **Give yourself TIME!** Plan on spending at least 2 hours studying outside of class for each hour we spend together in class or lab lecture. Do this every week, not just the week before the exam. Start early and it will be much easier later.
- If you consider yourself a poor test-taker, then you should complete and turn in all of the labs on time in order to pass the class. Also, utilize any practice exams or chapter reviews as they contain the same types of questions which you will encounter on the exams.
- Stay well rested and healthy. This is always a challenge in college, but do not neglect your basic needs. Poor sleep and diet have been shown to have a temporary negative impact on I.Q. Schedule study breaks as needed to keep up your mental health as well. Sometimes a night off is the right answer. Just don't make blowing off your studying a habit.
- As you listen, take notes, read, or work problems, try to keep an open mind, be curious, and think about the implications of the concepts and problems. Chemistry makes the world around us work and understanding why the world works will impress your friends at parties and help you grasp the material. The more connections you can make between the material in the book and the world around you, the more sense this class will make.

Tentative Lecture Schedule for Chem 25: Subject to Change

Week	Date	Section	Topics	
1	M 9/21 W 9/23	1.1-1.4 2.1-2.4	Classifying Matter and its Changes. Measurements Atomic Theory. Atomic Properties.	
2	M 9/28 W 9/30	2.5, 3.1-3.3 3.4-3.5	The Periodic Table. Ionic Compounds Naming ionic Compounds	
3	M 10/5 W 10/7	3.6-3.7, 4.1 4.2-4.4	Naming Acids and Bases. Percent Composition The mole: Calculations and Conversions. Solutions	
4	M 10/12 W 10/14	Exam 1 5.1-5.3	Chapter 1, 2, 3 What is a Chemical Reaction? How do we identify them? How do we write them?	
5	M 10/19 W 10/21	5.4-5.5 6.1-6.3	Classifying types of chemical reactions. Equations and Moles: Stoichiometry and Conversions	
6	M 10/26 W 10/28	6.4-6.5, 11.4 6.6-6.7, 7.1-7.2	Limiting Reactants and Percent Yield. Solutions. Changes in Energy and Heat. Energy and Atomic structure	
7	M 11/2 W 11/3	Exam 2 7.3-7.7	Ch 4, 5, 6 Modern Atomic Theory. Electron configuration and the Periodic Law. Periodic Trends	
8	M 11/9 W 11/11	8.1-8.3 8.4-8.5, 9.1-9.2	Ionic and Covalent bonds. Lewis Structures. Hydrocarbons. Molecular Geometry. Intro to Gasses	
9	M 11/16 W 11/18	9.2-9.5, 10.1 10.1-10.3, 13.1-13.2	The Ideal Gas Law. Gasses in Chemical Reactions Intro to Phase Changes. Phase Changes and Intermolecular forces Intro to Acids and Bases	
10	M 11/23 W 11/27	Exam 3 No Class	Ch 6, 7, 8, 9 Thanksgiving Holiday	
11	M 11/30 W 12/1	13.2-13.5, 11.5 14.1-2, 14.4 Review	Strong & Weak Acids. The pH Scale. Titrations Oxidation Numbers and Balancing Redox Reactions	
12	W		FINAL EXAM 7:00 AM-9:00 AM Cumulative	

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***Important Dates:

10/03 (Saturday): Last day to add classes

10/04 (Sunday): Last day to withdraw with no W on record 11/13 (Friday): Last day to withdraw with W on record

Tentative Lab Schedule for Chem 25: *Subject to Change* Fall 2020 De Anza College

Week	Date	Lab Topic	Due
1	M 9/21 W 9/23	Introduction, Safety, Check in	N/A
2	M 9/28 W 9/30	Lab 1-Chemical Safety	10/2
3	M 10/5 W 10/7	Lab 2-Matter and Phase Changes	10/9
4	M 10/12 W 10/14	Lab 3-Periodic Table of the Elements	10/16
5	M 10/19 W 10/21	Lab 4-Acids and Bases	10/23
6	M 10/26 W 10/28	Lab 5-Periodic Table	10/30
7	M 11/2 W 11/3	Lab 6-Atomic Structure	11/6
8	M 11/9 W 11/11	Lab 7-Basic Thermodynamics	11/13
9	M 11/16 W 11/18	Lab 8-Stoichiometric Calculations	11/20
10	M 11/23 W 11/27	Lab 9-Ionic and Covalent Bonds	11/27
11	M 11/30 W 12/1	Lab 10-Ideal Gas Law	12/4
12	-	-	No Lab – Finals Week

Student Learning Outcome(s):

^{*}Assess the fundamental concepts of modern atomic and molecular theory.

^{*}Evaluate the standard classes of chemical reactions.

^{*}Demonstrate a fundamental understanding of mathematical concepts pertaining to chemical experimentation and calculations.

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